

Name KEY Period A, C, G Date 10/10

USING MEASUREMENTS TO IDENTIFY MATTER - QUIZ REVIEW

PART A - ACCURACY, PRECISION, & PERCENT ERROR

1. Each of five students used the same ruler to measure the length of the same pencil. These data resulted: 15.33 cm, 15.34 cm, 15.33 cm, 15.33 cm, & 15.34 cm. The actual length of the pencil was 15.85 cm. Describe whether accuracy and precision are each good or poor for these measurements. Calculate the average, range, plus/minus amount, and the percent error for the set of measurements and then the percent error using the average length and the actual length of the pencil.

CALCULATIONS

Average: 15.33 cm
 Range: 0.01 cm
 \pm : ± 0.005 cm
 $\%E = (\pm \text{amount} \div \text{average}) \times 100$
 $\%E = 0.03\%$
 Very Precise

ACCURACY YES
 $\% \text{ ERROR} = \frac{15.85 - 15.33}{15.85} \times 100 = 3.3\%$
 $\%E = \left| \frac{\text{Actual} - \text{Experiment}}{\text{Actual}} \right| \times 100$

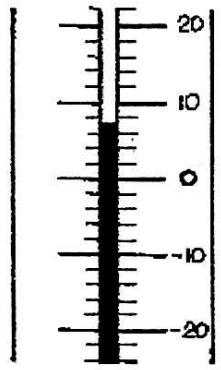
2. A chemistry student measured the boiling point of naphthalene ($C_{10}H_8$) at $231.0^\circ C$. What is the percent error for this measurement if the literature value is $217.9^\circ C$?

Accuracy $\rightarrow \%E = \frac{217.9 - 231.0}{217.9} \times 100 = 6.0\%$ Not very accurate

PART B - SIGNIFICANT FIGURES

3. 7°C

► Using the thermometer on the right, record the temperature in degrees Celsius. Show the correct number of significant figures.



Determine the number of significant figures in each of the following:

- 4. 3.57 m 3
- 5. 20.040 g 5
- 6. 0.004 m³ 1
- 7. 730 000 kg 2

Perform the following calculations and express your answers in the correct units and number of significant figures.

8. $(5.14 \text{ cm})(6.742 \text{ cm}) = \underline{34.7 \text{ cm}^2}$

9. $\frac{2.8 \text{ g}}{6.86 \times 10^{-3} \text{ L}} = \underline{410 \text{ g/L}}$

10.
$$\begin{array}{r} 45.68 \text{ g} \\ + 3.2 \text{ g} \\ \hline 48.88 \text{ g} \\ \hline 48.9 \text{ g} \end{array}$$

11.
$$\begin{array}{r} 1,423 \text{ mL} \\ - 40 \text{ mL} \\ \hline 1383 \text{ mL} \\ \hline 1380 \text{ mL} \end{array}$$

PART C - SCIENTIFIC NOTATION

Convert the following numbers into or out of scientific notation. Remember to keep the same number of significant figures.

12. 0.00003 cm $3 \times 10^{-5} \text{ cm}$

14. $1.05 \times 10^5 \text{ mm}$ 105,000 mm

13. 8,600,000 g $8.6 \times 10^6 \text{ g}$

15. $1.00 \times 10^{-3} \text{ m}$ 0.00100 m