| * | Read section 9.3 page 288-295 on pdf version of Ch. 9. |
|-------|--|
| | https://msrobbinspnhs.weebly.com/chemistry-2017-18.html (312 in Blue Text Book) |
| * | What are the 2 types of stoichiometry problems that deal with situations common in |
| | practical chemistry (the Lab)? Describe each. |
| | 1 |
| | · |
| | |
| | |
| | 2 |
| | |
| | |
| * | Describe what the limiting reactant is in a chemical reaction: |
| | |
| | |
| | |
| | |
| * | What is the excess reactant? |
| • | What is the excess reactain: |
| | |
| | |
| 5.00E | |
| * | EXTRA CREDIT: Page 292 in Modern Chemistry Limiting Reactants in a Cookie Recipe. |
| | Follow the procedure and answer the discussion questions. You must have the |
| | discussion questions completed for credit! (316 in Blue Text Book) |
| | |
| | Compare and contract the enotical and percent yield of products in a chemical position |
| ** | Compare and contrast theoretical and percent yield of products in a chemical reaction. |
| | |
| | |
| | |
| | |
| | |
| | |
| * | What is the formula for percent yield? |
| | |

Now Go to Page 295 and do #s 4-8 (pg. 321 #s 17-21 in Blue Text book)